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Observe how reactants and products interact in reversible reactions. The initial amount of each substance can be manipulated, as well as the pressure on the chamber. The amounts, concentrations, and partial pressures of each reactant and product can be tracked over time as the reaction proceeds toward equilibrium.

Equilibrium and Concentration. 2.4.3.c: perform K_c calculations involving the initial concentrations, the changes that occur in each substance, and the resulting equilibrium concentrations. Diffusion Equilibrium and Concentration. 2.4.3.d: predict the favourability of reactant or products in a reversible reaction, on the basis of the magnitude of the equilibrium constant. Equilibrium and Concentration Equilibrium and Pressure

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With Gizmos like Limiting Reactants and Equilibrium and Concentration, students are more able to visualize what is happening with the specific molecules in the reaction. Ms. Ringler explains, "In the Equilibrium and Concentration Gizmo, the students are able to see how some of the products of the original reaction break down to reform the reactants. Traditionally, this has been hard to represent on a larger scale so that they can compare quantities.

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